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~~Worksheet # 3~~

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Worksheet #1

Answers.

Stoichiometry

Worksheet #1

Answers 1. Given the

following equation: 2

$C_4H_{10} + 13 O_2 \rightarrow$

$8 CO_2 + 10 H_2O,$

show what the

following molar ratios

should be. a. C_4H_{10}

$/ O_2$ b. O_2 / CO_2 c.

O_2 / H_2O d. C_4H_{10}

$/ CO_2$ e. C_4H_{10} / H

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20 2. Given the
following equation: $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$
a.

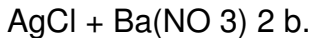
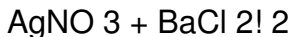
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Problems 1. Silver

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nitrate reacts with barium chloride to form silver chloride and barium nitrate. a.

Write and balance the chemical equation. 2



If 39.02 grams of barium chloride are reacted in an excess of silver nitrate, how many

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...

Dr. Rodriguez-Reyes
Chemistry Honors
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problems I. Answer
the questions for the
following reaction: $\text{Na (s)} + \text{Cl}_2 \rightarrow \text{NaCl (s)}$ 1.
How many moles of
Na are needed to
react with 12.50 mole

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Cl₂? 2. How many moles of Na are needed to produce 56.79 mole of NaCl?

3. How many moles of Cl₂ are needed to react with 33.50 mole Na? 4.

2018 Stoichiometry worksheet.docx - Dr Rodriguez-Reyes ...
Homework—Solving Stoichiometry

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Problems Name _____

ANSWERS _____ If
the statement is true,
write "true". If it is

false, change the
underlined word or
words to make it true.

Write your answer on
the line provided. _____

TRUE _____ 1. The
major types of
stoichiometry
problems are mass-
mass, mass-volume,

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and volume-volume.

Chemistry Homework—Solving Stoichiometry Problems 1

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many moles of
oxygen are consumed
when 96.7 moles of
hydrogen sulfide gas
are burned, producing
sulfur dioxide and

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water vapor in the process? 2. If 3.70×10^{23} molecules of oxygen react with excess benzene (C_6H_6), how many grams of water can be produced? 3.

Honors Chemistry
Practice Worksheet –
Stoichiometry
Honors Chemistry
Extra Stoichiometry

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Problems 1. Silver nitrate reacts with barium chloride to form silver chloride and barium nitrate. a. Write and balance the chemical equation. 2 $\text{AgNO}_3 + \text{BaCl}_2 \rightarrow 2 \text{AgCl} + \text{Ba}(\text{NO}_3)_2$ b. If 39.02 grams of barium chloride are reacted in an excess of silver nitrate, how many

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ty-Stoichiometry WS
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will count as your
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will submit a typed,
formal lab report,

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including all pre-lab and post-lab assignments. It will count as a formal grade worth 100 points. Alka-Seltzer is one of the world's best-known antacids. Its main function is to absorb excess stomach acid (HCl).

The Stoichiometry of
Alka Seltzer

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Mr. Christopherson /
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A student performs a double replacement reaction by mixing 500.0 ml of a 0.228 M solution of lithium carbonate with 370.0 ml of a 0.352 M solution of iron(III) chloride. The student

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collects the precipitate, and finds that 9.98 grams of precipitate are recovered. 1. Write the net ionic equation.

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 HO a).
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(stoichiometry) of 1
mol NO : 3 mol NO₂
(or it takes 3 mol NO₂
to make 1 mol
NO)... we can set up
the following
relationship (using
dimensional analysis):
1.2 moles NO₂ x 1

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mole NO / 3 moles
NO₂ = 0.4 moles NO
formed. 82). Write a
correctly balanced
equation for the
reaction taking place:
 $2\text{NO}(\text{g}) + \text{O}_2(\text{g}) \Rightarrow$
 $2\text{NO}_2(\text{g})$

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