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Activity A

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced

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to demonstrate
conservation of mass.

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the reason that they free

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like earning

appointments, giving directions towards clinic and answering a variety of other routine issues from patients. This function of handling calls can easily be delegated into a team experienced in taking and answering calls for the doctors.

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of other routine issues
from patients.

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This process is

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summarized by a
chemical equation. In
the Balancing Chemical
Equations Gizmo, look
at the floating molecules
below the initial

reaction: $\text{H}_2 + \text{O}_2$

H_2O . 1. How many
atoms are in a hydrogen
molecule (H_2)? 2 2.

How many atoms are in
an oxygen molecule (O_2)? 2 3. How many
hydrogen and oxygen

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atoms are in a water molecule (H_2O)? H: 2 O: 1 4. In general, what does a subscript (such as the “ 2 ” in H

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...

Check your answers for the first three formulas using the Gizmo.

$AgCl_3Cu_2$. Ag: 1

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Cl: 3. Cu:

2. Ba (AsO₄)₂ Ba:

1 As: 2. O:

8. 2018.

(NH₄)₃PO₄ N: 3

H: 12 P: 1...

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equations by changing
the coefficients of
reactants and products.

As the equation is
manipulated, the
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is shown as individual
atoms, histograms, or

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numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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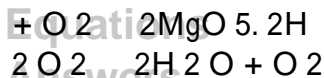
ANSWERS 1. $2\text{H}_2 +$

$\text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. 2Na

$+ \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. N_2

$\text{O}_4 \rightarrow 2\text{NO}_2$ 4. 2Mg

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Gizmo™, look at the floating molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. How many atoms are in a hydrogen molecule (H_2)? 2 How many atoms are in an oxygen molecule (O_2)? 2

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If you also get perplexed

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in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

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Objectives Students

will adjust

coefficients to balance a

chemical equation.

Classify chemical

equations as

combination (synthesis),

decomposition, single

replacement, or double

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replacement.

Vocabulary: coefficient,
combination,
compound,
decomposition, double
replacement, element,
molecule, product,
reactant, single
replacement ...

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topic is followed by discussions on the nature of intermolecular forces and the states of matter. This text further explores the statistics and dynamics of chemistry, including the study of equilibrium and kinetics. Other chapters cover the aspects of ionic equilibrium, acids and bases, and galvanic cells. The concluding

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chapters focus on a descriptive study of chemistry, such as the representative and transition elements, organic and nuclear chemistry, metals, polymers, and biochemistry. Teachers and undergraduate chemistry students will find this book of great value.

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The International
Hydrological Decade
(which ended in 1975)
led to a revival of
hydrological sciences to
a degree which, seen in
retrospect, is quite
spectacular. This
research programme
had strong government
support, no doubt due
to an increased
awareness of the role of
water for prosperous

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development. Since water quality is an essential ingredient in almost all water use, there was also a considerable interest in hydrochemistry during the Decade. As many concepts in classical hydrology had to be revised during and after the Decade there was also a need for revising hydrochemistry to align

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it with modern hydrology. A considerable input of fresh knowledge was also made in the recent past by chemists, particularly geochemists, invaluable for understanding the processes of mineralization of natural waters. With all this in mind it seems natural to try to

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assemble all the present knowledge of hydrochemistry into a book and integrate it with modern hydrology as far as possible, emphasizing the dynamic features of dissolved substances in natural waters.

Considering the role of water in nature for transfer of substances, this integration is

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Essential for proper understanding of processes in all related earth sciences. The arrangement of subjects in the book is as follows. After a short introductory chapter comes a chapter on elementary chemical principles of particular use in hydrochemistry.

Global warming

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continues to gain importance on the international agenda and calls for action are heightening. Yet, there is still controversy over what must be done and what is needed to proceed. Policy Implications of Greenhouse Warming describes the information necessary to make decisions about

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global warming
resulting from
atmospheric releases of
radiatively active trace
gases. The conclusions
and recommendations
include some
unexpected results. The
distinguished authoring
committee provides
specific advice for U.S.
policy and addresses the
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potential greenhouse warming. It offers a realistic view of gaps in the scientific understanding of greenhouse warming and how much effort and expense might be required to produce definitive answers. The book presents methods for assessing options to reduce emissions of greenhouse gases into

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the atmosphere, offset emissions, and assist humans and unmanaged systems of plants and animals to adjust to the consequences of global warming.

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serves as an essential companion to the science curriculum supplement, STEPS to STEM. As students learn key concepts in the seven “big ideas” in this program (Electricity & Magnetism; Air & Flight; Water & Weather; Plants & Animals; Earth & Space; Matter & Motion; Light &

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Sound), they record their ideas, plans, and evidence. There is ample space for students to keep track of their observations and findings, as well as a section to reflect upon the use of “ Science and Engineering Practices ” as set forth in the Next Generation Science Standards (NGSS).

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notebook is reflective of the behavior of scientists. One of the pillars of the Nature of Science is that scientists must document their work to publish their research results; it is a necessary part of the scientific enterprise. This is important because STEPS to STEM is a program for young scientists who

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learn within a
community of scientists.
Helping students to
think and act like
scientists is a critical
feature of this program.
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need to keep a written
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teacher. Teachers

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should also model writing in science to help instill a sense of purpose and pride in using and maintaining a Sci-Book. Lastly, students' documentation can serve as a valuable form of authentic assessment; teachers can utilize Sci-Books to monitor the learning process and the development of science

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